

TESTING PROCEDURE



No. TM 02	Titrimetry	Revision No.: 0
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1. Apparatus

- Buret and stand
- Volumetric flask
- Pipette
- Measuring cylinders
- Conical flask

2. Reagent

- 0.1N iodine VS
- Water
- 2 N sulfuric acid
- Starch TS

3. Procedure

Sample : 400 mg of Ascorbic Acid

Titrimetric system :

1. Mode : Direct titration
2. Titrant : 0.1N iodine VS
3. Endpoint detection : Visual
4. Blank : 100 mL of water and 20 mL of 2N sulfuric acid. Add 3 mL of starch TS

Analysis:

Dissolve the *Sample* in a mixture of 100 mL of water and 25 mL of 2N sulfuric acid. Add 3 mL of starch TS, and titrate immediately with *Titrant* until a persistent violet-blue color is obtained. Calculate the percentage of ascorbic acid (C₆H₈O₆) in the portion of Ascorbic Acid taken:

$$\text{Result} = [(V-B) \times N \times F \times 100]/W$$

- V = sample titrant volume (mL)
B = blank titrant volume (mL)
N = titrant normality (mEq/mL)
F = equivalent factor, 88.06 mg/mEq
W = weight of Sample (mg)