A. IDENTIFICATION BY TLC

1. SCOPE AND FIELD OF APPLICATION
The method describes the identification of hydroquinone in cosmetic products.

2. PRINCIPLE
Hydroquinone is identified by thin layer chromatography (TLC).

3. REAGENTS
All reagents must be of analytical grade, except for Hydroquinone RS
3.1. Absolute ethanol
3.2. n-Hexane
3.3. Acetone
3.4. Toluene
3.5. Glacial acetic acid
3.6. Phosphomolybdic acid
3.7. Silver nitrate
3.8. Ammonium hydroxide 25% (w/w)
3.9. Developing solvents:
   3.9.1. System n°1: n-Hexane/Acetone, 3:2 (v/v)
   3.9.2. System n°2: Toluene/Glacial acetic acid, 8:2 (v/v)
3.10. Reference Standard (RS): Hydroquinone RS
3.11. Spray reagents
   3.11.1. To a 5 % (w/v) aqueous solution of silver nitrate, add ammonium hydroxide until the precipitate which forms is dissolved.
   Warning: the solution becomes explosively unstable on standing and should be discarded after use.
   3.11.2. 5 % (w/v) solution of phosphomolybdic acid in ethanol

4. APPARATUS
Normal laboratory equipment, and:
4.1. TLC plates, ready for use: silica gel 60 F254 20 cm x 20 cm. Layer thickness 0.25 mm
4.2. Ultrasonic bath
4.3. UV lamp, 254 nm

5. PROCEDURE
5.1. Preparation of the Sample Solution
   5.1.1. Weigh about 1.5 g of sample into a 25 mL beaker.
   5.1.2. Add gradually 15 mL of ethanol 96 % (v/v), and mix.
   5.1.3. Transfer into a 25 mL volumetric flask.
   5.1.4. Homogenize in an ultrasonic bath for 10 minutes, and then cool the flask to room temperature.
   5.1.5. Add ethanol 96 % (v/v) to volume, and mix.
   5.1.6. Put in an ice bath until separation of fats occurs (indication time: 10 minutes).
   5.1.7. Filter through a paper filter.
5.2.1 Weigh about 0.05 g of Hydroquinone RS into a 25 mL volumetric flask.
5.2.2 Add 5 ml of ethanol 96 % (v/v), and shake to dissolve it.
5.2.3 Add ethanol 96 % (v/v) to volume, and mix.

5.3. Thin Layer Chromatography (TLC)
5.3.1 Activate the plates for 10 minutes at 100 °C.
5.3.2 Saturate two chromatographic tanks with both developing solvents.
5.3.3 Deposit on both plates 20 µL of
- sample solution
- reference solution

Notes:
- spots may be duplicated,
- spiked sample solution (prepared by mixing one mL of reference solution with one mL of sample solution) may be spotted in parallel.
- If the spot is too large, spotted be adjusted (reduced), i.e. 2 - 5 µL

5.3.4 Develop in a dark at room temperature until the solvent front has migrated 15 cm from the start.
5.3.5 Remove the plates and allow to dry at room temperature.
5.3.6 Detection
5.3.6.1 Observe the plate under UV light at 254 nm, and mark the position of the spots.
5.3.6.2 Spray the plate with:
- silver nitrate reagent, the formation of black spot or
- phosphomolybdic acid reagent, and heat the plate to approximately 100°C for about 10 minutes, when necessary, the formation of blue spot.

6. IDENTIFICATION
6.1. Calculate the Rf value for each spot:
- their Rf values,
- the colour of the spots under UV radiation,
- and the colours of the spots after visualization with the spray reagent.
6.2. Perform the HPLC described in the following section (B), and compare the retention times obtained for the sample peak with that for the standard solution.
6.3. Combine the results from TLC and HPLC to identify the presence of hydroquinone.

7. REMARKS
Method validation information

<table>
<thead>
<tr>
<th>No</th>
<th>Developing Solvents</th>
<th>Rf</th>
<th>LOD (µg/g)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>n-hexane/acetone, 3:2</td>
<td>0.32</td>
<td>100</td>
</tr>
<tr>
<td>2.</td>
<td>toluene/glacial acetic acid, 8:2</td>
<td>0.20</td>
<td>150</td>
</tr>
</tbody>
</table>
B. IDENTIFICATION AND DETERMINATION BY HPLC

1. SCOPE AND FIELD OF APPLICATION
This method specifies a procedure for the identification and determination of hydroquinone in cosmetic products.

2. PRINCIPLE
The sample is extracted with a water/methanol mixture under gentle heating to melt any lipid material. Determination of the hydroquinone in the resulting solution is performed by reversed phase liquid chromatography with UV detection.

3. REAGENTS
All reagents must be of analytical quality and suitable for HPLC where appropriate, except for Hydroquinone RS. Water used must be distilled water, or water of at least equivalent purity.

3.1 Methanol (HPLC grade)
3.2 Mobile phase: water/methanol mixture 45:55 (V/V). Mix 55 volumes of methanol and 45 volumes of water.
3.3 Reference Standard (RS) : Hydroquinone RS

4. APPARATUS
Normal laboratory equipment and:
4.1 Water bath, capable of maintaining a temperature of 60 °C
4.2 High-performance liquid chromatograph with Photo Diode Array (PDA) detector and 20-µL injection loop
4.3 Analytical column : Stainless steel chromatographic column, length 250 mm, internal diameter 4.6 mm, packed with ODS-C18, particle size 5 µm, or equivalent.
4.4 Filter paper, diameter 90 mm, Schleicher and Schull, Weisshand No 5892, or equivalent (Whatman n° 1)
4.5 Vortex mixer
4.6 0.45 µm membrane filter (HVLP or equivalent)
4.7 Laboratory centrifuge

5. PROCEDURE
5.1 Preparation of the Standard Solution
5.1.1 Weigh accurately 0.0125 g (wref) of Hydroquinone RS into a 50 mL volumetric flask.
5.1.2 Add 25 mL of mobile phase and shake to dissolve it.
5.1.3 Add mobile phase to volume, and mix.
5.1.4 Pipette 5 mL this solution into a 50 mL volumetric flask.
5.1.5 Dilute and add mobile phase to volume, then mix.

Notes:
- dilution may be adapted to sample concentration.
- these solutions must be freshly prepared. Amber (low actinic) glass may be used.

5.2 Preparation of the Sample Solution
5.2.1 Weigh accurately 1 ± 0.1 g of sample (wspl) into a 25 mL beaker.
5.2.2 Add gradually 25 mL of mobile phase, and mix until homogeneous.
5.2.3 Transfer into a 50 mL volumetric flask.
5.2.4 Vortex for about 1 minute.
5.2.5 Place the flask in water bath at 60°C for 15 minutes, then cool the flask to room temperature.
5.2.6 Add mobile phase to volume, and mix.
5.2.7 Filter the clear solution through a 0.45 μm membrane filter (when necessary, centrifuge first for 10 min).
5.2.8 Perform identification and determination of filtrate by HPLC within less than 24 hours.

5.3 High Performance Liquid Chromatography

5.3.1 Conditions

<table>
<thead>
<tr>
<th>Column</th>
<th>ODS-C18, 5 μm, 250 x 4.6 mm or equivalent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mobile phase</td>
<td>Water - Metanol (45:55)</td>
</tr>
<tr>
<td>Flow rate</td>
<td>1.0 mL/minute</td>
</tr>
<tr>
<td>PDA detector</td>
<td>292 nm</td>
</tr>
<tr>
<td>Inject volume</td>
<td>20 μL</td>
</tr>
</tbody>
</table>

5.3.2 Inject 20 μl of the standard and sample solutions obtained as described in section 5.1 - 5.2 and record the chromatograms. Measure the peak areas. Compare the chromatograms obtained for sample and standard solutions.

5.3.3 System suitability

Inject 20 μL of the reference solution and record the chromatogram. Inject 6 times to ensure that a constant peak area is obtained. [CV = (standard deviation/mean) x 100, should be less than 2%].

6. CALCULATION

Use the areas of the hydroquinone peaks to calculate the concentration of the hydroquinone in the sample. Calculate the hydroquinone concentration in the sample, as a percentage by mass, (xi) using the formula:

\[
\text{Hydroquinone, } \% \text{ w/w} = \frac{a_{\text{sample}} \times w_{\text{std}} (g)}{a_{\text{std}} \times w_{\text{sample}} (g)} \times D \times 100
\]

where:
\(a_{\text{sample}}\) = peak area of sample
\(a_{\text{std}}\) = peak area of standard
\(w_{\text{sample}}\) = weight of sample
\(w_{\text{std}}\) = weight of standard
\(D\) = dilution factor = 0.1 in the conditions above

7. REMARKS

7.1 Method validation information

<table>
<thead>
<tr>
<th>No.</th>
<th>Parameters of Validation</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.</td>
<td>Retention time</td>
<td>3 minutes</td>
</tr>
<tr>
<td>2.</td>
<td>Precision</td>
<td>1.63 %</td>
</tr>
<tr>
<td>3.</td>
<td>Accuracy</td>
<td>(97.0 - 98.7) %</td>
</tr>
<tr>
<td>4.</td>
<td>LOD</td>
<td>7.5 µg/g</td>
</tr>
<tr>
<td>5.</td>
<td>LOQ</td>
<td>25 µg/g</td>
</tr>
</tbody>
</table>
7.2 Chromatogram of Hydroquinone

**Harmonised method:**

- Issued by the chemical analysis group at the harmonization workshop in Kuala-Lumpur, on September 13\(^{th}\) to 17\(^{th}\), 2004
- Approved by the harmonization workshop delegates workshop in Kuala-Lumpur, on September 13\(^{th}\) to 17\(^{th}\), 2004,
- Modified after the Jakarta training, Nov 22\(^{nd}\) to 26\(^{th}\), 2004
- Modified and approved after the Brunei workshop, Aug 30\(^{th}\) to 31\(^{st}\), 2005
- Modified and approved after the final review in Singapore, Nov 30\(^{th}\) to Dec 2\(^{nd}\), 2005
- Modified and approved after the 1\(^{st}\) ACTLC Meeting in Solo (Indonesia), Nov 20\(^{nd}\), 2012